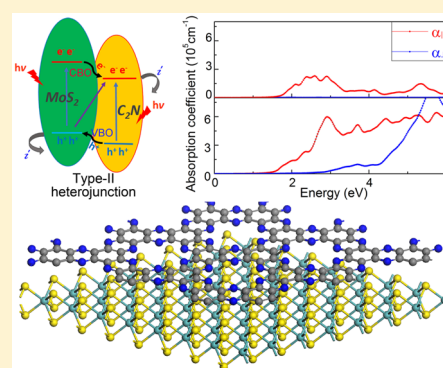


Tunable Structural, Electronic, and Optical Properties of Layered Two-Dimensional C₂N and MoS₂ van der Waals Heterostructure as Photovoltaic Material

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S Supporting Information

ABSTRACT: The nitrogenated porous two-dimensional (2D) material C₂N has been successfully synthesized using a simple wet-chemical reaction, which provides a high-performance way to produce such 2D materials with novel electronic and optical properties. In this work, density functional theory (DFT) calculations were performed to investigate the structural, electronic, and optical properties of the layered C₂N/MoS₂ van der Waals (vdW) heterojunction. The C₂N/MoS₂ heterojunction was found to have a direct band gap of 1.30 eV and to present the typical type-II heterojunction feature, facilitating the effective separation of photogenerated electrons and holes. The calculated band alignment and enhanced optical absorption suggest that the C₂N/MoS₂ heterojunction should exhibit good light-harvesting properties. The vertical strain can effectively tune the electronic properties and optical absorption of the C₂N/MoS₂ heterojunction by changing the interaction between the p_z orbital of C₂N and the d_{z²} orbital of MoS₂. The moderate band gap, well-separated photogenerated electrons and holes, and enhanced visible-light absorption indicate that the C₂N/MoS₂ heterojunction is a potential photovoltaic structure for solar energy.



1. INTRODUCTION

Two-dimensional (2D) materials have attracted much attention because of their novel electronic, thermal, and optoelectronic properties.^{1–12} In addition to graphene,^{1,2} many other 2D systems have also been subjected to extensive studies in recent years, including hexagonal boron nitride (*h*-BN) monolayers,^{3–5} transition-metal dichalcogenides (TMDs),^{6–10} C₃N₄,¹¹ and C₂N.¹² MoS₂, as a representative TMD (MX₂),^{6–10} has a direct band gap (2.0 eV) and a high carrier mobility in the limit of the single monolayer,⁸ making it an important photocatalytic and photovoltaic material.^{9,10} Very recently, a C₂N monolayer with uniform pore and nitrogen-atom distributions was synthesized successfully and found to be a semiconductor with a direct band gap of 1.96 eV.¹² The unique electronic properties of C₂N and MoS₂ could be complementary to the use of graphene and *h*-BN in application areas^{6–10,12–22} such as catalysis,^{13,14} electronics,^{15–18} molecular sieves,^{19,20} photocatalysis,^{12,21,22} and photovoltaics.^{8,12} However, a significant challenge still remains for the use of C₂N and MoS₂ in photocatalysis and photovoltaic cells: The charge distributions of their valence-band maximum (VBM) and conduction-band minimum (CBM) states are not well-separated in space, resulting in significantly reduced light-

absorbing efficiency because of the fast recombination of photogenerated electrons and holes.^{6,12} The construction of heterostructures by stacking different 2D materials on top of each other with interlayer van der Waals (vdW) interactions is a potential way to overcome this weakness.^{23,24}

Notably, vdW heterojunctions have the advantage of utilizing the qualities of the stacked 2D materials and have thus been widely investigated;^{23–51} examples include graphene/*h*-BN,^{34,35} graphene/MoS₂,^{38–41} MX₂/M'X'₂,^{42–51} and C₃N₄/MoS₂ heterojunctions.^{23,24} Among the MX₂/M'X'₂ vdW heterojunctions, MoSe₂/MoS₂,^{42,49} WSe₂/MoS₂,^{43,44,46,50} MoSe₂/WSe₂,⁴⁷ and WS₂/MoS₂⁴⁸ heterojunctions present type-II band alignments, but their electronic and optical properties are closely related to the stacking order,⁵¹ which does not favor flexible control in experiments.⁴⁸ As potential light-harvesting materials, C₂N and MoS₂ monolayers have nearly the same band gaps,^{8,12} whereas their work functions differ.¹² Therefore, the C₂N/MoS₂ vdW heterojunction is expected to have a reduced direct band gap that is beneficial for photoabsorption.

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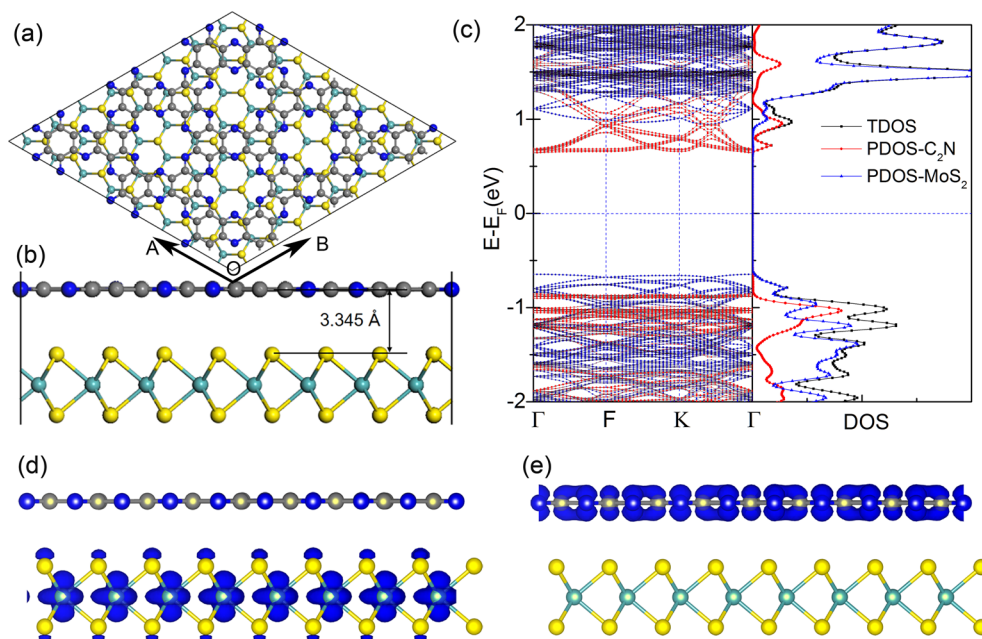


Figure 1. (a,b) Top and side views of the optimized C₂N/MoS₂ heterojunction. (c) Band structure and PDOS of the C₂N/MoS₂ heterojunction. The sizes of the red and blue dots represent the weights of C₂N and MoS₂, respectively. The black, red, and blue lines present the total density of states of the C₂N/MoS₂ heterojunction and the PDOS of C₂N and MoS₂, respectively. The Fermi level (E_F) was set to 0 eV. (d,e) Charge densities of the (d) VBM and (e) CBM of the C₂N/MoS₂ heterojunction. The isovalue was set to 0.03 e Å⁻³.

It was therefore interesting to investigate whether this heterojunction could show appropriate band alignment to inhibit electron–hole recombination.

In this work, we investigated the C₂N/MoS₂ vdW heterojunction as a potential photovoltaic structure using first-principles calculations. The calculated electronic band structure showed that the C₂N/MoS₂ heterojunction exhibits type-II band-alignment characteristics with a direct band gap of ~1.3 eV. The type-II band alignment facilitates the separation of photogenerated electrons and holes, as the photogenerated electrons can easily transfer from the MoS₂ layer to the C₂N layer and vice versa for the photogenerated holes. The calculated optical spectra showed enhanced absorption of the visible and ultraviolet light, especially for the in-plane component, implying that the C₂N/MoS₂ vdW heterojunction should exhibit promising photovoltaic properties. The vertical compressive strain was found to effectively tune the band-gap energy and enhance the optical absorption of the C₂N/MoS₂ heterojunction. The C₂N/MoS₂ heterojunction still exhibits the type-II band alignment under certain compressive strains.

2. COMPUTATIONAL DETAILS

Our calculations were performed using density functional theory (DFT) with the projector augmented-wave method,⁵² as implemented in the Vienna ab initio simulation package (VASP).⁵³ The generalized gradient approximation (GGA) was used for the exchange–correlation functional with the Perdew–Burke–Ernzerhof (PBE) parametrization.⁵⁴ To accurately describe the vdW interaction, we employed the empirical correction method presented by Grimme (DFT-D2),⁵⁵ which has proven to be reliable for describing long-range vdW interactions. As a benchmark, we calculated the distance between bilayer graphene to be 3.25 Å with the PBE + D2 method, with a corresponding binding energy of about 25 meV per carbon atom. These results are in good agreement with previous experimental and theoretical studies.^{56,57}

The vacuum region was set to at least 15 Å to avoid virtual interactions between adjacent images. The cutoff energy for plane-wave expansion was 520 eV. For the isolated C₂N and MoS₂ monolayers, the first Brillouin zone integration was performed using Γ -centered $6 \times 6 \times 1$ and $10 \times 10 \times 1$ Monkhorst–Pack k -point samplings, respectively. For the C₂N/MoS₂ heterojunction, we used Γ -centered $2 \times 2 \times 1$ and $3 \times 3 \times 1$ k -point samplings for geometry optimization and electronic structure calculations, respectively. All of the structures were relaxed without any symmetry restriction until the total forces acting on all atoms were less than 0.01 eV/Å.

To investigate the optical absorption properties of the C₂N/MoS₂ heterojunction, we calculated the frequency-dependent dielectric function [$\epsilon(E)$] and evaluated the absorption coefficient [$\alpha(E)$] using the expression⁵⁸

$$\alpha(E) = \frac{4\pi E}{hc} \left[\frac{(\epsilon_1^2 + \epsilon_2^2)^{1/2} - \epsilon_1}{2} \right]^{1/2}$$

where ϵ_1 and ϵ_2 represent the real and imaginary parts, respectively, of the frequency-dependent complex dielectric function.⁵⁸ Because of the structural anisotropy, anisotropic optical absorption was expected for the heterojunction, with $\alpha(E)^{xx} = \alpha(E)^{yy} \neq \alpha(E)^{zz}$. Correspondingly, the absorption coefficient was divided into two parts: parallel (α_{\parallel}) and perpendicular (α_{\perp}) to the plane of the C₂N monolayer.

3. RESULTS AND DISCUSSION

Before investigating the C₂N/MoS₂ heterojunction, we first explored the structural and electronic properties of isolated C₂N and MoS₂ monolayers. The optimized lattice parameters for the MoS₂ and C₂N monolayers are 3.183 and 8.318 Å, respectively, in good agreement with the experimental measurements and previous theoretical results.^{7,12} The C₂N monolayer (Figure S1a) is composed of evenly distributed pores and nitrogen atoms, with C–N distances of 1.336 Å and

C–C distances of 1.427 and 1.467 Å, and all of the atoms are nearly in the same plane. The band structure and partial density of states (PDOS) calculated with the PBE parametrization (see Figure S1b,c) indicate that C₂N is a semiconductor with a direct band gap of 1.66 eV. For the MoS₂ monolayer, we considered only the most stable honeycomb structure (H-MoS₂).¹² The optimized Mo–S distance in the H-MoS₂ monolayer is about 2.413 Å, and the calculated band structure shows that it is a semiconductor with a direct gap of 1.67 eV. These observations are consistent with those of previous reports.^{8,12,19,20,59–61}

3.1. Geometric Structure of the C₂N/MoS₂ Heterojunction. In our calculations, the C₂N/MoS₂ heterojunction was constructed based on a supercell model, in which 3 × 3 × 1 C₂N and 8 × 8 × 1 MoS₂ cells were used to match with each other. The supercell used consisted of 354 atoms, including 108 C, 54 N, 64 Mo, and 128 S atoms. The lattice mismatch for this supercell was only 2.3%.

Top and side views of the optimized C₂N/MoS₂ heterojunction are shown in panels a and b, respectively, of Figure 1. In the optimized geometry, the C₂N layer maintains its original planar structure without visible distortion, and MoS₂ also maintains its original plane of hexagonally arranged molybdenum atoms sandwiched between two planes of hexagonally arranged sulfur atoms. The equilibrium distance between the C₂N and MoS₂ layers was calculated to be 3.345 Å, close to the distance (3.280 Å) in a C₂N bilayer with AB stacking.⁵⁹ We calculated the binding energy (E_b) of the heterojunction according to the equation

$$E_b = E_{C_2N/MoS_2} - E_{C_2N} - E_{MoS_2}$$

where E_{C_2N/MoS_2} , E_{C_2N} , and E_{MoS_2} are the total energies of the relaxed C₂N/MoS₂ heterojunction, isolated C₂N, and MoS₂ monolayers, respectively. A negative E_b value of −19.98 meV (per atom) was obtained, which indicates that the formation of the C₂N/MoS₂ heterojunction is energetically favorable.

3.2. Electronic Structure and Band Alignment. We next explored the electronic structure of the C₂N/MoS₂ heterojunction. The calculated band structure, shown in Figure 1c, indicates that the C₂N/MoS₂ heterojunction is a semiconductor with a direct band gap of 1.30 eV, which is consistent with the results of PBE-D3 and self-consistent vdW calculations and smaller than those of C₂N (1.66 eV) and MoS₂ (1.67 eV). Hence, electron excitation from the valence band (VB) to the conduction band (CB) would be easier for the C₂N/MoS₂ heterojunction under visible-light irradiation. From the calculated PDOS in Figure 1c, it can be seen that the VBM is mainly provided by the states from the MoS₂ layer whereas the CBM is dominated by the states from the C₂N layer. The valence-band offset (VBO) and conduction-band offset (CBO) between the C₂N and MoS₂ layers were calculated to be about 0.20 and 0.32 eV, respectively. These band-alignment properties suggest that C₂N and MoS₂ form a type-II heterojunction. As illustrated in Figure 2a, the CB and VB of the MoS₂ layer are higher in energy than the corresponding bands of the C₂N layer. When the C₂N/MoS₂ heterojunction is irradiated with light, the electrons can be photoexcited from the valence bands of both the C₂N and MoS₂ sheets. Because of the CBO, the photogenerated electrons in the MoS₂ layer can be transferred to the CB of the C₂N layer. Meanwhile, the VBO promotes the transfer of photogenerated holes from the VB of the C₂N layer to that of the MoS₂ layer. Therefore, both the VBO and CBO

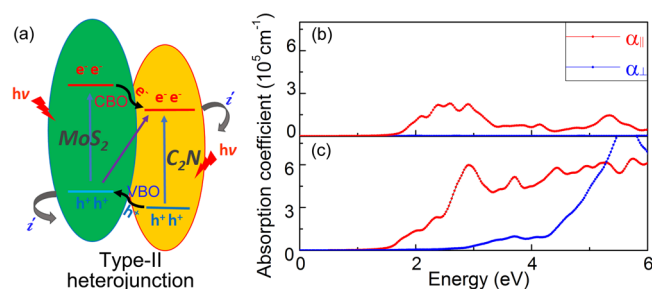


Figure 2. (a) Schematic representation of the band alignment including both VBO and CBO for the C₂N/MoS₂ heterojunction. (b,c) Absorption coefficients of the (b) C₂N monolayer (3 × 3 supercell) and (c) C₂N/MoS₂ heterojunction. α_{\parallel} and α_{\perp} represent the optical absorption coefficients parallel and normal to the plane of the C₂N monolayer, respectively.

promote the redistribution of photogenerated electrons and holes. In addition, the calculated charge densities of the VBM and CBM, shown in Figure 1d,e, indicate that the VBM is made up of the d orbital of MoS₂ whereas the CBM is made up of the p orbital of C₂N, consistent with our detailed PDOS results. The full separation of the charge distributions of the VBM and CBM in the C₂N/MoS₂ heterojunction further promotes the separation of photogenerated electrons and holes, reducing their recombination. More accurate calculations using the Heyd–Scuseria–Ernzerhof (HSE06) hybrid functional (provided in the Supporting Information) confirmed the above results.

3.3. Optical Properties. To investigate the combined effects of C₂N and MoS₂, we calculated the optical properties of both the C₂N monolayer and the C₂N/MoS₂ heterojunction. The obtained optical absorption coefficients with the polarization vector parallel (α_{\parallel}) and normal (α_{\perp}) to the C₂N plane are shown in Figure 2b for the C₂N monolayer. Clearly, the optical absorption is dominated by the α_{\parallel} component, whereas the contribution of the α_{\perp} component is negligible. The absorption edge of C₂N is about 1.6 eV, in good agreement with its band gap (1.66 eV). Four absorption peaks were calculated for the α_{\parallel} component at 2.10, 2.39, 2.59, and 2.90 eV with intensities as large as $2.0 \times 10^5 \text{ cm}^{-1}$, which originate from different interband transitions between the VB and CB of C₂N. These optical behaviors reflect well the ability of the C₂N monolayer to be used in solar cells to absorb visible light (1.61 eV < E < 3.10 eV). The calculated optical absorption coefficients of the C₂N/MoS₂ heterojunction are shown in Figure 2c. The absorption edge is about 1.3 eV, smaller than that of C₂N. The red shift of the absorption edge is due to the fact that, in the C₂N/MoS₂ heterojunction, the photogenerated electrons can be directly excited from the VB of the MoS₂ to the CB of the C₂N. Furthermore, in comparison with the isolated C₂N monolayer, an obvious enhancement of the optical absorption was found for the C₂N/MoS₂ heterojunction. In particular, for the α_{\parallel} component, four significant absorption peaks appeared at 2.02, 2.33, 2.91, and 3.70 eV with large intensities of 1.38×10^5 , 2.28×10^5 , 5.97×10^5 , and $5.14 \times 10^5 \text{ cm}^{-1}$, respectively. Two absorption peaks at 3.70 and 5.70 eV were also calculated for the α_{\perp} component with intensities as high as 1.0×10^5 and $8.6 \times 10^5 \text{ cm}^{-1}$, respectively, indicating improved out-of-plane optical absorption compared to that of the isolated C₂N monolayer. In a word, the C₂N/MoS₂ heterojunction was found to exhibit obviously enhanced optical absorption for both visible and ultraviolet light.

3.4. Effect of the Vertical Strain. The vertical strain has proven to be effective in modulating the electronic properties of vdW heterojunctions by changing the interactions between different sheets.^{17,62–64} In experiments, one can effectively control the vertical strain applied to the vdW heterojunction by the strain device using diamond anvil cells.⁶⁵ Here, we systematically investigated the strain effect in the C₂N/MoS₂ vdW heterojunction. We defined the vertical strain as $\epsilon = d_0 - d$, where d_0 and d are the equilibrium and actual distances, respectively, between the C₂N and MoS₂ sheets. As shown in Figure 3e, with increasing distance between the C₂N and MoS₂

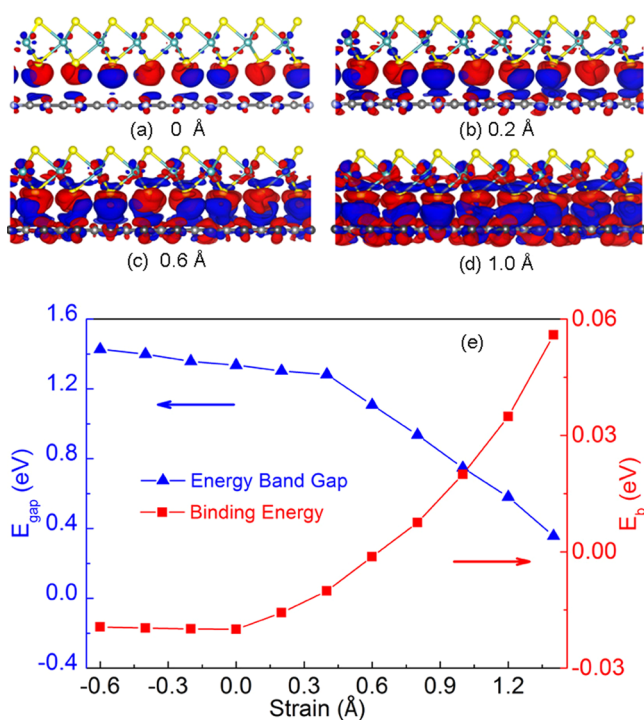


Figure 3. (a–d) Charge density differences of the C₂N/MoS₂ heterojunction with vertical strains: (a) 0, (b) 0.2, (c) 0.6, and (d) 1.0 Å. Blue and red represent charge accumulation and depletion, respectively. (e) Evolution of the band-gap energy E_{gap} and binding energy E_{b} as a function of the applied strain. The iso-value was set to 0.02 e \AA^{-3} .

layers, the binding energy was found to rise slowly. For $\epsilon = -0.2, -0.4,$ and -0.6 \AA , the corresponding E_{b} values were $-19.88, -19.65,$ and -19.37 meV , respectively. For the compressed strain, the binding energy increased quickly and changed to 0 eV when the distance between the C₂N and MoS₂ sheets approached 2.745 \AA ($\epsilon = 0.6 \text{ \AA}$).

The change in the interaction between the C₂N and MoS₂ layers should be reflected by the intensity of charge transfer between them. To explore the charge-transfer process, the charge density difference of the C₂N/MoS₂ heterojunction was calculated, as shown in Figure 3a–d for selected strains. For $\epsilon = 0 \text{ \AA}$, a charge redistribution appeared at the interface, and about $0.36 e$ of charge was determined to be transferred from the MoS₂ layer to the C₂N layer by integrating the differential electron density curve.⁶⁶ As the distance between the C₂N and MoS₂ sheets decreased, the charge transfer obviously intensified as a result of the enhanced interlayer interaction. Thus, the electronic behavior of the C₂N/MoS₂ heterojunction is expected to be well tuned by compressive strain.

The calculated band-gap energy (E_{gap}) of the C₂N/MoS₂ heterojunction as a function of the applied strain is shown in Figure 3e, and the evolution of the corresponding PDOS is shown in Figure 4. With increasing compressive strain, the band

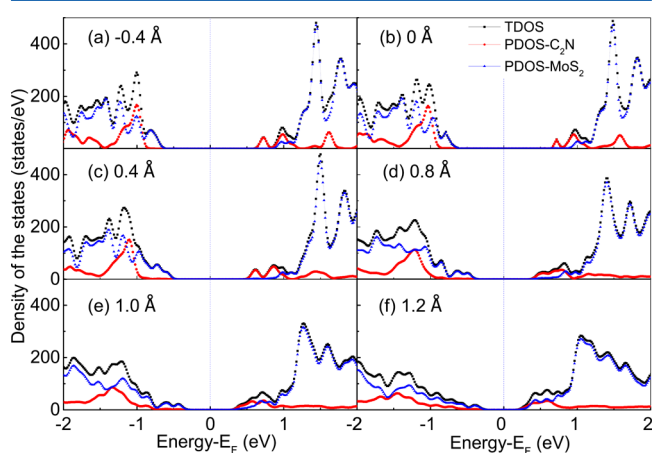


Figure 4. PDOS of the C₂N/MoS₂ heterojunction with vertical strains of (a) -0.4 , (b) 0 , (c) 0.4 , (d) 0.8 , (e) 1.0 , and (f) 1.2 \AA .

gap decreased monotonically, as a result of the enhanced interaction between the p_z orbital of C₂N and the d_z^2 orbital of MoS₂. For ϵ values from -0.6 to 0.3 \AA , E_{gap} was found to follow a linear relationship with strain, namely, $E_{\text{gap}} = 1.336 - 0.147\epsilon$, whereas for ϵ values in the range 0.4 – 1.4 \AA , a different linear relationship of $E_{\text{gap}} = 1.675 - 0.929\epsilon$ was observed. It is noteworthy that, for strains of $\epsilon < 1.0 \text{ \AA}$, the C₂N/MoS₂ heterojunction was always found to present type-II band alignment. As shown in Figure 4f, when the strain increased to 1.2 \AA , only C₂N contributed to the CBM whereas both C₂N and MoS₂ contributed to the VBM, thus giving a band alignment that deviated from the type-II band feature. For even larger strains of $\epsilon > 1.90 \text{ \AA}$, the C₂N/MoS₂ heterojunction was found to undergo a semiconductor-to-metal transition, implying tunable conductive and transport properties of this heterojunction.

Finally, we examined the optical properties of the C₂N/MoS₂ heterojunction under vertical strain. The absorption coefficients calculated for selected strains are shown in Figure 5. With increasing compressive strain, both the α_{\parallel} and α_{\perp} components were found to be significantly red-shifted, consistent with the decrease in the band-gap energy discussed above. There was obvious optical absorption for infrared and visible light. The effective modulation of optical absorption by vertical strain implies wide potential applications of C₂N/MoS₂ heterojunctions in devices for photodetection and photovoltaic structures used in solar cells.

4. SUMMARY

In summary, the structural, electronic, and optical properties of the C₂N/MoS₂ vdW heterojunction were systematically investigated using first-principles calculations. The C₂N/MoS₂ heterojunction presents typical type-II band alignment with a direct band gap of 1.3 eV , which facilitates the effective separation of photogenerated electrons and holes. The calculated band alignment and enhanced optical absorption indicate that the C₂N/MoS₂ vdW heterojunction exhibits good light-harvesting properties. The electronic and optical properties of the C₂N/MoS₂ heterojunction were found to be

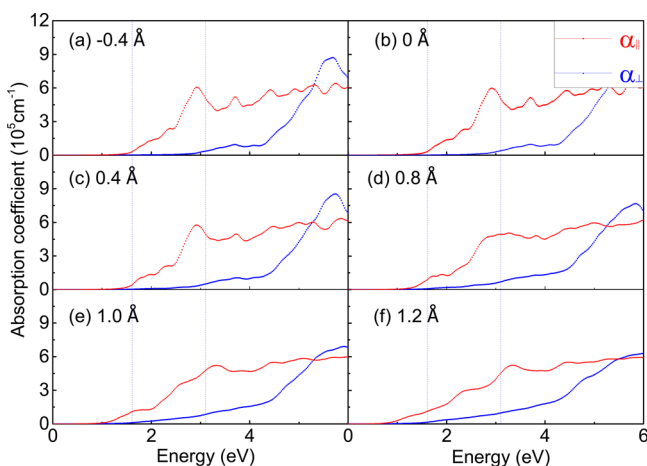


Figure 5. Absorption coefficient of the C_2N/MoS_2 heterojunction with vertical strains (a) -0.4 , (b) 0 , (c) 0.4 , (d) 0.8 , (e) 1.0 , and (f) 1.2 Å.

effectively tuned by vertical strain through changes in the interaction between the p_z orbital of C_2N and the d_z^2 orbital of MoS_2 . The C_2N/MoS_2 heterojunction with a moderate band gap, well-separated photogenerated electrons and holes, and enhanced visible-light absorption has great potential for use as a photovoltaic heterojunction in solar cells.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jpcc.6b12681.

Information on materials, band structure and PDOS of the C_2N monolayer, band structure of the C_2N/MoS_2 heterojunction calculated by PBE-D3, PDOS of the heterojunction calculated by HSE06, and differential electron density curve of the heterojunction (PDF)

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Notes

The authors declare no competing financial interest.

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